

Data collection

Rigaku AFC-5R diffractometer 1341 reflections with $I > 2\sigma(I)$
 $\omega-2\theta$ scans $R_{\text{int}} = 0.035$
 Absorption correction: $\theta_{\text{max}} = 70^\circ$
 ψ scans (North *et al.*, 1968) $h = -5 \rightarrow 8$
 $T_{\text{min}} = 0.661$, $T_{\text{max}} = 0.947$ $k = 0 \rightarrow 39$
 3541 measured reflections $l = -3 \rightarrow 6$
 1736 independent reflections 3 standard reflections
 every 150 reflections
 intensity decay: 3.8%

Refinement

Refinement on F $w = 1/[\sigma^2(F_o) + 0.00141|F_o|^2]$
 $R = 0.05$ $(\Delta/\sigma)_{\text{max}} = 0.0004$
 $wR = 0.070$ $\Delta\rho_{\text{max}} = 0.15 \text{ e } \text{\AA}^{-3}$
 $S = 1.380$ $\Delta\rho_{\text{min}} = -0.25 \text{ e } \text{\AA}^{-3}$
 1341 reflections Extinction correction: none
 118 parameters Scattering factors from
 H-atom parameters not International Tables for
 refined Crystallography (Vol. C)

Table 1. Selected geometric parameters (\AA , $^\circ$)

O1—C1	1.417 (3)	C5—C6	1.519 (3)
O2—C11	1.421 (3)	C6—C7	1.512 (3)
C1—C2	1.502 (3)	C7—C8	1.511 (3)
C2—C3	1.517 (3)	C8—C9	1.516 (3)
C3—C4	1.518 (3)	C9—C10	1.511 (3)
C4—C5	1.508 (3)	C10—C11	1.500 (3)
O1—C1—C2	112.7 (2)	C6—C7—C8	113.6 (2)
C1—C2—C3	114.6 (2)	C7—C8—C9	115.1 (2)
C2—C3—C4	112.7 (2)	C8—C9—C10	112.8 (2)
C3—C4—C5	114.7 (2)	C9—C10—C11	114.4 (2)
C4—C5—C6	113.7 (2)	O2—C11—C10	109.0 (2)
C5—C6—C7	114.8 (2)		
O1—C1—C2—C3	-63.3 (3)	O2—C11—C10—C9	179.3 (2)

Table 2. Hydrogen-bonding geometry (\AA , $^\circ$)

D—H...A	D—H	H...A	D...A	D—H...A
O2—H2O...O1 ⁱ	0.96	1.76	2.710 (2)	172.0
O1—H1O...O2 ⁱⁱ	0.97	1.81	2.775 (3)	173.9

Symmetry codes: (i) $2 - x, \frac{1}{2} + y, -\frac{1}{2} - z$; (ii) $\frac{3}{2} - x, 1 - y, z - \frac{1}{2}$.

All non-H atoms were refined anisotropically by full-matrix least-squares methods. All H atoms including hydroxyl H atoms were fixed in idealized positions.

Data collection: *MSC/AFC Diffractometer Control Software* (Molecular Structure Corporation, 1992). Cell refinement: *MSC/AFC Diffractometer Control Software*. Data reduction: *TEXSAN* (Molecular Structure Corporation, 1995). Program(s) used to solve structure: *SAPI91* (Fan, 1991). Program(s) used to refine structure: *TEXSAN*. Software used to prepare material for publication: *TEXSAN*.

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Supplementary data for this paper are available from the IUCr electronic archives (Reference: OB1004). Services for accessing these data are described at the back of the journal.

References

- Fan, H.-F. (1991). *SAPI91. Structure Analysis Programs with Intelligent Control*. Rigaku Corporation, Tokyo, Japan.
 Johnson, C. K. (1976). *ORTEPII*. Report ORNL-5138. Oak Ridge National Laboratory, Tennessee, USA.
 Kobayashi, H. & Nakamura, N. (1995). *Cryst. Res. Technol.* **30**, 495–500.
 Kobayashi, H., Yamamoto, T. & Nakamura, N. (1995). *Cryst. Res. Technol.* **30**, 375–380.
 Kupke, S., Seidei, I., Szulzewsky, K., Steger, U. & Steger, E. (1981). *Cryst. Res. Technol.* **16**, 349–356.
 Molecular Structure Corporation (1992). *MSC/AFC Diffractometer Control Software*. MSC, 3200 Research Forest Drive, The Woodlands, TX 77381, USA.
 Molecular Structure Corporation (1995). *TEXSAN. Single Crystal Structure Analysis Software*. Version 1.7. MSC, 3200 Research Forest Drive, The Woodlands, TX 77381, USA.
 Nakamura, N. & Setodoi, S. (1997). *Acta Cryst.* **C53**, 1883–1885.
 Nakamura, N., Tanihara, Y. & Takayama, T. (1997). *Acta Cryst.* **C53**, 253–255.
 Nakamura, N. & Yamamoto, T. (1994). *Acta Cryst.* **C50**, 946–948.
 North, A. C. T., Phillips, D. C. & Mathews, F. S. (1968). *Acta Cryst.* **A24**, 351–359.
 Seto, T. (1962). *Mem. Coll. Sci. Univ. Kyoto Ser. A*, **30**, 89–107.
 Watanabe, A. (1961). *Bull. Chem. Soc. Jpn.* **34**, 1728–1734.

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Tautomerism in a 2,4-dichlorobenzene-sulfonamide derived from 2-amino-4-(2,5-dimethoxyphenyl)thiazole

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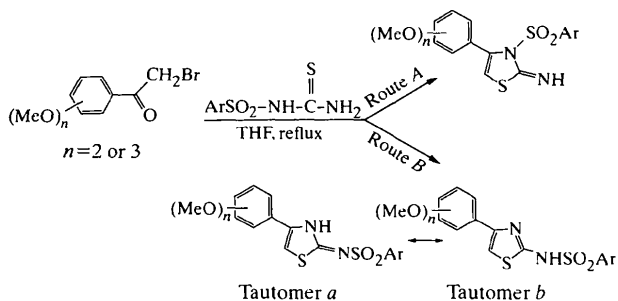
(Received 19 October 1998; accepted 18 December 1998)

Abstract

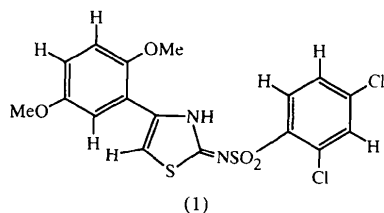
2,4-Dichloro-*N*-[2,3-dihydro-4-(2,5-dimethoxyphenyl)thiazol-2-ylidene]benzenesulfonamide, $\text{C}_{17}\text{H}_{14}\text{Cl}_2\text{N}_2\text{O}_4\text{S}_2$, crystallizes in space group $P\bar{1}$ with two independent molecules. The structural parameters do not indicate a tautomeric equilibrium but a single imino form. The main differences between the two crystalline forms lie in the intramolecular hydrogen bonding and the relative orientation of the methoxy groups. Attractive intermolecular interactions occur and are responsible for the crystalline cohesion.

Comment

With the objective of designing potential antifungal agents, we have focused our efforts on the preparation of new thiazole derivatives incorporating both an arylsulfonamide moiety and a polyoxygenated phenyl group (Beuchet *et al.*, 1999). These compounds were obtained by the Hantzsch condensation between an α -bromo ketone and an appropriate arylsulfonamide thiourea, according to the scheme below.



The present X-ray study was undertaken in order to elucidate the cyclization pathway (route A or route B) and to determine the prominent form in the tautomeric equilibrium (*a* or *b*). The structure determination of the product, (1), obtained from 1-(2,5-dimethoxyphenyl)-2-bromoethanone and *N*-(aminothioxomethyl)-2,4-dichlorobenzene sulfonamide is reported. The main



feature lies in the presence of two independent molecules (designated I and II; Fig. 1) contained in each asymmetric unit. The arylsulfonamide group is linked unambiguously to the exocyclic N atom (N12 or N62), thus indicating the cyclization to occur *via* route B. A comparison of intramolecular bond lengths and angles (Table 1) found for the five-membered ring indicates certain similarities between the two conformations. Both structures are characterized by the inequality of the two C—S bonds. The distances C9—S10 [1.747 (3) Å] and C59—S60 [1.732 (3) Å] are longer than C11—S10 [1.727 (3) Å] and C61—S60 [1.720 (4) Å]. A similar situation has been encountered with the crystal structures of various benzo- and naphthothiazoles (Fehlmann, 1970; Ekstrand & van der Helm, 1977). According to these reports, we can consider these extended C—S distances in our heterocyclic system as pure single bonds. For both molecules I and II, analysis of C—N distances

around C9 and C59 shows that the exocyclic double bond [mean value 1.312 (4) Å] is shorter than its intracyclic counterpart [mean value 1.341 (4) Å]. These data are in agreement with those found for 2-iminothiazole derivatives (Cohen-Addad *et al.*, 1981; Cohen-Addad, 1982; Kalcheva *et al.*, 1993). Furthermore, each of molecules I or II possesses an H atom located near the endocyclic N atom, as shown from the final difference Fourier maps. These observations are indicative of the prominence of the imino tautomer *a* in the crystalline state.

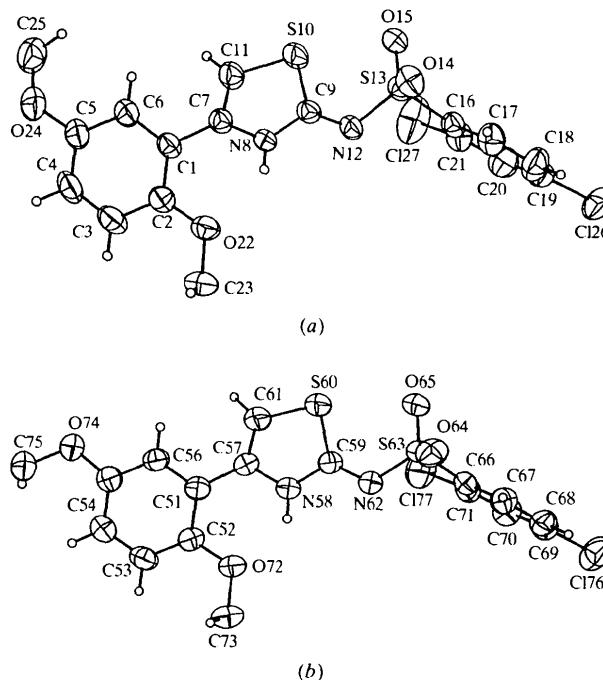


Fig. 1. The molecular structure and atomic numbering scheme of (a) molecule I and (b) molecule II, showing 50% probability displacement ellipsoids for all non-H atoms. H atoms have been set artificially small.

Inspection of the dihedral angles relative to the dimethoxyphenyl ring reveals some differences between the two molecules. In the first molecule (I), the methoxy groups are on opposite sides of the molecule, whereas they are on the same side in molecule II. It is interesting to note that, in spite of the absence of steric hindrance, the dimethoxyphenyl moiety of molecule I exists with a non-planar conformation, generated by a 36.8 (6)° rotation of the methoxy O24—C25 group around the C5—O24 bond. Some interactions of special interest, including both intramolecular and intermolecular interaction, are given in Table 2. As a main structural feature for molecule II, the O72 atom participates in an intramolecular hydrogen bond with the heterocyclic N—H group. The N58...O72 contact [2.627 (4) Å] falls short

of the corresponding van der Waals radii sum by 0.32 Å (Emsley, 1989). On the other hand, in the structure of I, the distance N8...O22 is increased significantly [2.808(4) Å] and thus indicates a weaker intramolecular interaction than in molecule II. This diversity in the strength of hydrogen bonding reflects the variation in the distribution of angles between the heterocyclic and the aromatic rings. Comparison of torsion angles (Table 1) shows that dimethoxyphenyl and dihydrothiazolyl moieties are less twisted in conformation II [15.0(1)°] than in conformation I [39.1(1)°]. Furthermore, a close intramolecular approach is observed between S60 and O65 whose distance [2.723(4) Å] is shorter than the calculated sum of the van der Waals radii for O and S (3.25 Å; Emsley, 1989). This non-bonding attractive interaction between S and O atoms is in accordance with previously reported data for a benzoylimino-1,3-thiazolidine derivative (Cohen-Addad *et al.*, 1981). The hydrogen bond involving N8 and O14ⁱ [symmetry code: (i) $-x, 1-y, 1-z$] is among the molecular distances of particular interest given in Table 2. The crystalline cohesion is achieved by numerous van der Waals contacts and also by C—H...O attractive interactions.

Experimental

Crystals for X-ray structural analysis were grown by recrystallization of the title compound from an acetone–heptane solution.

Crystal data

C₁₇H₁₄Cl₂N₂O₄S₂

$M_r = 445.32$

Triclinic

$P\bar{1}$

$a = 12.369(1) \text{ \AA}$

$b = 12.998(1) \text{ \AA}$

$c = 13.390(1) \text{ \AA}$

$\alpha = 63.54(1)^\circ$

$\beta = 88.97(1)^\circ$

$\gamma = 79.89(1)^\circ$

$V = 1892.9(3) \text{ \AA}^3$

$Z = 4$

$D_x = 1.568 \text{ Mg m}^{-3}$

D_m not measured

Data collection

Enraf–Nonius CAD-4 diffractometer

ω - 2θ scans

Absorption correction: ψ scan (North *et al.*, 1968)

$T_{\min} = 0.486$, $T_{\max} = 0.583$

6620 measured reflections

6303 independent reflections

Cu $K\alpha$ radiation

$\lambda = 1.54178 \text{ \AA}$

Cell parameters from 25 reflections

$\theta = 3.64$ – 64.94°

$\mu = 5.393 \text{ mm}^{-1}$

$T = 296(2) \text{ K}$

Prism

$0.15 \times 0.12 \times 0.10 \text{ mm}$

White

4512 reflections with $I > 2\sigma(I)$

$R_{\text{int}} = 0.013$

$\theta_{\max} = 64.94^\circ$

$h = -14 \rightarrow 14$

$k = 0 \rightarrow 15$

$l = -13 \rightarrow 14$

2 standard reflections

frequency: 90 min

intensity decay: none

Refinement

Refinement on F^2

$R[F^2 > 2\sigma(F^2)] = 0.041$

$wR(F^2) = 0.101$

$S = 1.043$

6303 reflections

488 parameters

H atoms not refined

$w = 1/[\sigma^2(F_o^2) + (0.0782P)^2 + 2.2037P]$

where $P = (F_o^2 + 2F_c^2)/3$

$(\Delta/\sigma)_{\max} = -0.029$

$\Delta\rho_{\max} = 0.318 \text{ e \AA}^{-3}$

$\Delta\rho_{\min} = -0.301 \text{ e \AA}^{-3}$

Extinction correction:

SHELXL93

Extinction coefficient:

0.0145(15)

Scattering factors from

International Tables for Crystallography (Vol. C)

Table 1. Selected geometric parameters (Å, °)

C1—C7	1.475(4)	C51—C57	1.473(5)
C7—C11	1.336(4)	C57—C61	1.343(5)
C7—N8	1.390(4)	C57—N58	1.386(4)
N8—C9	1.341(4)	N58—C59	1.341(4)
C9—N12	1.313(4)	C59—N62	1.311(4)
C9—S10	1.747(3)	C59—S60	1.732(3)
S10—C11	1.727(3)	S60—C61	1.720(4)
N12—S13	1.591(3)	N62—S63	1.593(3)
S13—O14	1.438(2)	S63—O64	1.437(3)
S13—O15	1.440(2)	S63—O65	1.441(3)
S13—C16	1.775(3)	S63—C66	1.769(4)
C11—C7—N8	111.2(3)	C61—C57—N58	109.8(3)
N8—C7—C11	122.1(3)	N58—C57—C51	121.9(3)
C9—N8—C7	116.8(2)	C59—N58—C57	117.2(3)
N12—C9—N8	120.5(3)	N62—C59—N58	120.3(3)
N12—C9—S10	131.0(2)	N62—C59—S60	130.9(3)
N8—C9—S10	108.5(2)	N58—C59—S60	108.9(2)
C11—S10—C9	91.0(2)	C61—S60—C59	90.7(2)
C7—C11—S10	112.5(3)	C57—C61—S60	113.3(3)
C9—N12—S13	120.8(2)	C59—N62—S63	120.5(2)
O14—S13—O15	116.14(15)	O64—S63—O65	116.4(2)
O14—S13—N12	108.23(14)	O64—S63—N62	108.8(2)
O15—S13—N12	113.59(14)	O65—S63—N62	117.2(2)
O14—S13—C16	105.28(15)	O64—S63—C66	105.7(2)
O15—S13—C16	108.09(14)	O65—S63—C66	108.0(2)
N12—S13—C16	104.56(14)	N62—S63—C66	104.3(2)
C1—C2—O22—C23	174.3(3)	C51—C52—O72—C73	172.7(3)
C6—C5—O24—C25	36.8(6)	C54—C55—O74—C75	−5.7(5)

Table 2. Selected geometric parameters (Å, °)

Hydrogen bonds			
D—H...A	D...A	H...A	D—H...A
N58—H58...O72	2.627(5)	2.031(5)	125.6(4)
N8—H8...O14 ⁱ	2.994(4)	2.168(4)	161.4(4)
N8—H8...O22	2.809(4)	2.423(4)	107.9(3)

C—H...O interactions

C—H...O	C...O	H...O	C—H...O
C11—H11...O64 ⁱⁱ	3.158(6)	2.402(6)	138.3(5)
C73 ⁱⁱⁱ —H73C ⁱⁱⁱ ...O15	3.289(6)	2.644(6)	124.9(5)

van der Waals contacts

S60...O65	2.723(4)	O64...C11 ^{iv}	3.156(6)
O65...C177	3.210(4)	C126...C23 ^v	3.554(6)
O14...O22 ⁱ	2.962(4)	S60...C75 ^{vi}	3.625(6)
N62...C177	3.307(4)	C126...C126 ^{vii}	3.285(2)
C61...C127 ⁱⁱⁱ	3.584(5)		

Symmetry codes: (i) $-x, 1-y, 1-z$; (ii) $x, y-1, z$; (iii) $1-x, 1-y, 1-z$; (iv) $x, 1+y, z$; (v) $1+x, y, z$; (vi) $1-x, 2-y, 2-z$; (vii) $1-x, 1-y, 2-z$.

Data collection: CAD-4 Software (Enraf–Nonius, 1989). Cell refinement: CAD-4 Software. Data reduction: CAD-4 Software. Program(s) used to solve structure: SHELXS86 (Sheldrick, 1990). Program(s) used to refine structure: SHELXL93 (Sheldrick, 1993).

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Supplementary data for this paper are available from the IUCr electronic archives (Reference: GS1030). Services for accessing these data are described at the back of the journal.

References

- Beuchet, P., Varache-Lembége, M., Neveu, A., Léger, J.-M., Larrouture, S., Deffieux, G. & Nuhrich, A. (1999). *Eur. J. Med. Chem.* In the press.
- Cohen-Addad, C. (1982). *Acta Cryst.* B38, 1753–1757.
- Cohen-Addad, C., Savariault, J.-M. & Lehmann, M. S. (1981). *Acta Cryst.* B37, 1703–1706.
- Ekstrand, J. D. & van der Helm, D. (1977). *Acta Cryst.* B33, 1012–1016.
- Emsley, J. (1989). In *The Elements*. Oxford: Clarendon Press.
- Enraf-Nonius (1989). *CAD-4 Software*. Version 5.0. Enraf-Nonius, Delft, The Netherlands.
- Fehlmann, M. (1970). *Acta Cryst.* B26, 1736–1741.
- Kalcheva, V., Tosheva, M. & Hadjieva, P. (1993). *Ann. Chem.* pp. 1319–1322.
- North, A. C. T., Phillips, D. C. & Mathews, F. S. (1968). *Acta Cryst.* A24, 351–359.
- Sheldrick, G. M. (1990). *Acta Cryst.* A46, 467–473.
- Sheldrick, G. M. (1993). *SHELXL93. Program for the Refinement of Crystal Structures*. University of Göttingen, Germany.

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(+)-(2*R*-*cis*)-Dimethyl(10-bromo-2,3,3*a*,8-tetrahydrodibenz[*c*,*f*]isoxazolo[2,3-*a*]azepin-2-ylmethyl)amine†

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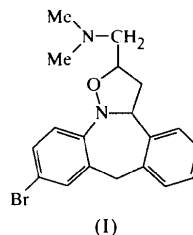
Abstract

The title compound, C₁₉H₂₁BrN₂O, is a central active serotonin 5-HT_{2c} antagonist with some H₁ affinity. The (6,7,6)-tricyclic moiety is asymmetrically folded with a dihedral angle of 124.3(1)° between the aromatic planes. The fused tetrahydroisoxazole ring adopts a

conformation halfway between that of a twist and an envelope.

Comment

The structure of the title compound, (I), has been determined as part of our studies on serotonin antagonists (Peeters *et al.*, 1995, and references cited therein). The



compound is a central active serotonin 5-HT_{2c} antagonist with some H₁ affinity. *In vivo* the compound antagonizes anxiety symptoms (Meert, 1998).

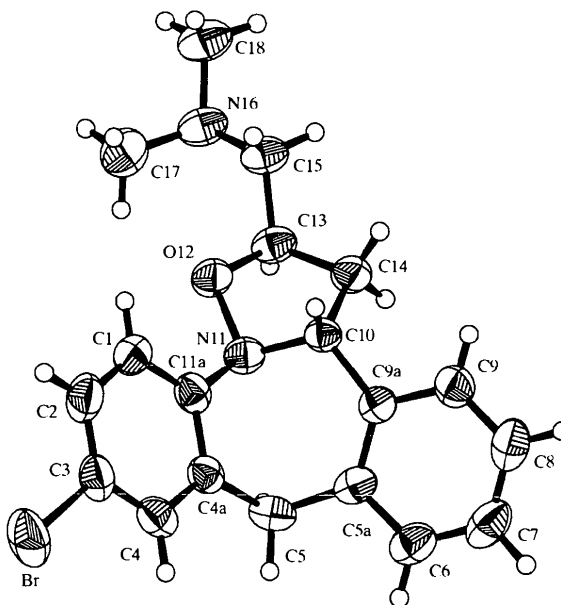


Fig. 1. Perspective view of the title compound with atomic numbering scheme. Displacement ellipsoids are drawn at the 50% probability level.

The conformation and the atomic numbering scheme are shown in Fig. 1. The (6,7,6)-tricyclic moiety folds asymmetrically about a line through C5 and N11. The dihedral angle between the least-squares planes of the aromatic rings is 124.3(1)°. The bond lengths do not show outstanding features. Apart from the endocyclic angles of the seven-membered ring, the bond angles have normal values. The endocyclic angles facing the non-bromo-substituted benzo ring are markedly enlarged. This enlarging and the

† Internal code of the Janssen Research Foundation: R110580.